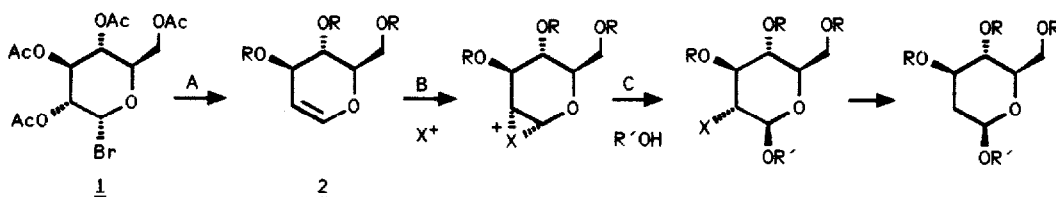


AN EASY SYNTHESIS OF 2'-DEOXY- β -DISACCHARIDES¹

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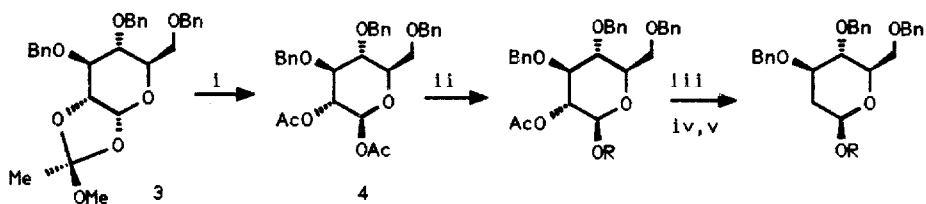
Abstract: 1,2-Di-O-acetyl-3,4,6-tri-O-benzyl- β -D-glucopyranose and analogs are excellent β -glycosyl donors in the presence of trimethylsilyltriflate. The resulting disaccharides are easily converted into the corresponding 2'-deoxy- β -disaccharides, opening a practical route to this important class of natural molecules.

2-Deoxy- β -D-glycosides are sugar components of various natural products, such as the orthosomycin group² of antibiotics and a series of cytostatics including chromomycin A₃³, olivomycin A⁴, and mithramycin⁵. Reported methods for the selective preparation of such glycosides often benefit⁶ from the directing effect of a temporary neighboring group at C-2, which is subsequently removed under mild conditions, leaving behind the desired 2-deoxy- β -glycoside as shown in the following scheme :



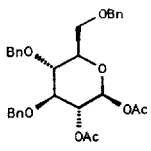
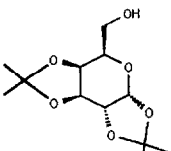
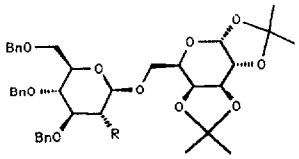
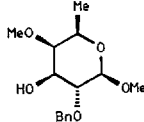
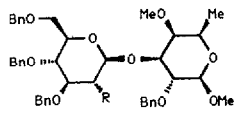
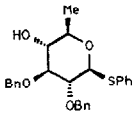
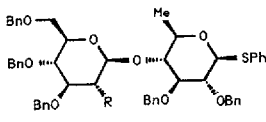
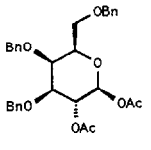
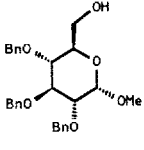
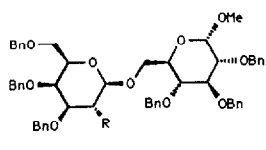
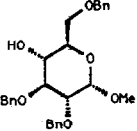
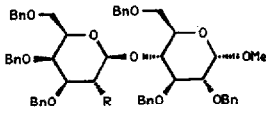
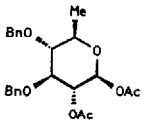
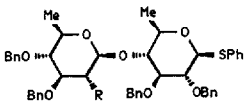
Analysis of this current circuitous strategy reveals a shortcoming. The initial and appropriate orientation of the participating O-acetyl group at C-2 in compound **1** is deliberately lost during the formation of glucal **2** so that the now problematic issue of the glycosylation results from the rather critical stereochemical outcome of steps B and C⁷.

We would like to report on a *fully stereoselective, general and high-yielding synthesis of 2'-deoxy- β -disaccharides*, where the stereochemical outcome is directly governed by the natural orientation of the O-acetyl group at C-2 as shown in the following scheme :



Reagents: i) dry CH_3COOH , 1 h, 20°C; ii) ROH (1 equiv.),

TABLE 1

GLYCOSYL DONOR	GLYCOSYL ACCEPTOR	DISACCHARIDE ¹² (Yield %)
		
<u>4</u>		<u>11</u> R = OAc (85) ↪ <u>12</u> R = H (70)
<u>4</u>		
	<u>7</u> ¹⁹	<u>13</u> R = OAc (78) ↪ <u>14</u> R = H (82)
<u>4</u>		
	<u>8</u> ²¹	<u>15</u> R = OAc (75) ↪ <u>16</u> R = H (82)
		
<u>5</u>	<u>9</u> ²³	<u>17</u> R = OAc (81) ↪ <u>18</u> R = H (95)
<u>5</u>		
	<u>10</u> ²⁴	<u>19</u> R = OAc (90) ↪ <u>20</u> R = H (77)
	<u>8</u>	
<u>6</u>		<u>21</u> R = OAc (80) ↪ <u>22</u> R = H (80)

CH₂Cl₂, 4 Å MS, TMSOTf (1 equiv.), 20 min, -20°C; *iii*) MeOH, NaOMe, 1 h, 20°C; *iv*) NaH, imidazole, CS₂, ICH₃, THF, 1 h, 20°C; *v*) Bu₃SnH (10 equiv.), AIBN, toluene, 3 h, reflux⁹.

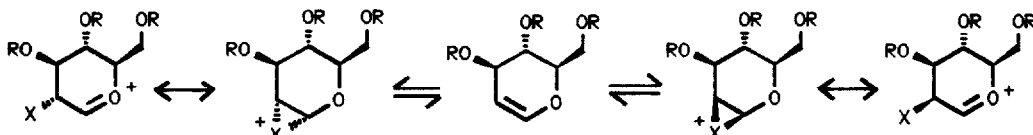
Although easily available benzylated orthoesters were directly used as glycosyl donors^{10,11}, yields and selectivities were not always high. Known **3**¹⁰ was thus converted into the corresponding 1,2-trans-di-*O*-acetyl derivative **4** (85%)¹². In the presence of trimethylsilyltriflate¹³, **4** was consistently an excellent β-glycosyl donor at -20°C, as shown in Table I, which also includes the results with 1,2-trans-di-*O*-acetyl derivatives **5**¹⁴ and **6**¹⁵.

A somewhat similar approach to the problem has been the selective deoxygenation at C-2' of a natural β-linked disaccharide¹⁸, but the regioselective manipulation of protecting groups in disaccharides is not always straightforward.

In conclusion, the availability of benzylated β-acetates of types **4**, **5** and **6** opens the route to a simple, probably general, efficient and stereospecific preparation of a variety of 2'-deoxy-β-disaccharides. A limitation of this approach lies in the impossibility of using acylated glycosyl acceptors, inasmuch as regiospecific deoxygenation at C-2' would be rather problematic. In order to enlarge the scope of this strategy we thus developed another related approach, whereby the protection of either the glycosyl donor or acceptor with *O*-acyl was made feasible. It is presented in the following letter.

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- 1 Part of this work was presented at the 4th European Carbohydrate Symposium, Darmstadt (Germany), A-121, 1987.
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- 7 The kinetic diastereofacial selective formation of an onium intermediate does not insure the stereoselective formation of the β-glycoside, in that the following equilibration may take place⁸



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- 12 All new compounds gave satisfactory microanalytical and spectral data. Values of $[\alpha]_D$ and δ_H were measured for CHCl_3 and CDCl_3 solutions, at 20° and 25°C, respectively.
- 4**: $[\alpha]_D$ +25.5°; δ_H : 5.64 (1 H, d, $J_{1,2}$ 8.5 Hz, H-1).
- Selected physical data for key disaccharidic compounds:
- 11**: $[\alpha]_D$ -34°; δ_H : 5.54 (1 H, d, $J_{1,2}$ 5.0 Hz, H-1), 4.46 (1 H, d, $J_{1',2'}$ 8.0 Hz, H-1'), 5.03 (1 H, dd, $J_{2',3'}$ 9.0 Hz, H-2').
- 12**: $[\alpha]_D$ -47°; δ_H : 5.61 (1 H, d, $J_{1,2}$ 5.0 Hz, H-1), 4.54 (1 H, dd, $J_{1',2'}$ eq 2.0, $J_{1',2'}$ ax 9.5 Hz, H-1').
- 13**: m.p. 136°C (ethanol), $[\alpha]_D$ +12°; δ_H : 5.01 (1 H, d, $J_{1',2'}$ 8.0 Hz, H-1'), 4.17 (1 H, d, $J_{1,2}$ 7.5 Hz, H-1).
- 14**: m.p. 95°C (CH_2Cl_2 -hexane), $[\alpha]_D$ +2°; δ_H : 4.68 (1 H, dd, $J_{1',2'}$ eq 2.0, $J_{1',2'}$ ax 10 Hz, H-1'), 4.22 (1 H, d, $J_{1,2}$ 7.5 Hz, H-1).
- 15**: m.p. 129-130°C (ethanol), $[\alpha]_D$ +22°; δ_H : 5.04 (1 H, dd, $J_{1',2'}$ 8.5, $J_{2',3'}$ 9.5 Hz, H-2').
- 16**: m.p. 128-129°C (ethanol), $[\alpha]_D$ +10°; δ_H : 2.35 (1 H, ddd, $J_{1',2'}$ eq 2.5, $J_{2'}$ eq, $J_{2'}$ ax 12.5, $J_{2'}$ eq, $J_{3'}$ 4.5 Hz, H-2'eq), 1.63 (1 H, ddd, $J_{1',2'}$ ax 10, $J_{2'}$ ax, $J_{3'}$ 9.5 Hz, H-2'ax).
- 17**: m.p. 117°C (AcOEt-hexane), $[\alpha]_D$ +10°; δ_H : 4.42 (1 H, d, $J_{1',2'}$ 8 Hz, H-1').
- 18**: m.p. 100°C (ethanol), $[\alpha]_D$ +7°; δ_H : 4.24 (1 H, dd, $J_{1',2'}$ ax 9.5, $J_{1',2'}$ eq 2.0 Hz, H-1').
- 19**: $[\alpha]_D$ +9°; δ_H : 5.33 (1 H, dd, $J_{1',2'}$ 8.0, $J_{2',3'}$ 10 Hz, H-2').
- 20**: $[\alpha]_D$ -3°, δ_H : 4.75 (1 H, dd, $J_{1',2'}$ eq 2.0, $J_{1',2'}$ ax 10 Hz, H-1').
- 21**: m.p. 150-151°C (ethanol), $[\alpha]_D$ +25°; δ_H (C_6D_6): 4.75 (1 H, d, $J_{1,2}$ 9.5 Hz, H-1), 4.55 (1 H, d, $J_{1',2'}$ 8.0 Hz, H-1').
- 22**: m.p. 136-138°C (ethanol), $[\alpha]_D$ +6°, δ_H (C_6D_6): 4.63 (1 H, d, $J_{1,2}$ 9.5 Hz, H-1), 4.36 (1 H, dd, $J_{1',2'}$ eq 2.0, $J_{1',2'}$ ax 10 Hz, H-1').
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- 14 **5** (84%); $[\alpha]_D$ +23°; δ_H : 5.62 (1 H, d, $J_{1,2}$ 8.5 Hz, H-1). It was prepared from known 3,4,6-tri-*O*-benzyl-1,2-*O*-(1-methoxyethylidene)- α -*D*-galactopyranose¹⁵.
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- 16 **6**: m.p. 112-113°C (ethanol), $[\alpha]_D$ +24°; δ_H : 5.66 (1 H, d, $J_{1,2}$ 8.5 Hz, H-1) was prepared from 2,3,4-tri-*O*-acetyl-6-deoxy- α -*D*-glucopyranosyl bromide¹⁷ by conventional methods.
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